

Synthesis of Grevillins, Novel Pyrandione Pigments of Fungi. Biogenetic Interrelationships between Grevillins, Pulvinic Acids, Terphenylquinones and Xylerythrins

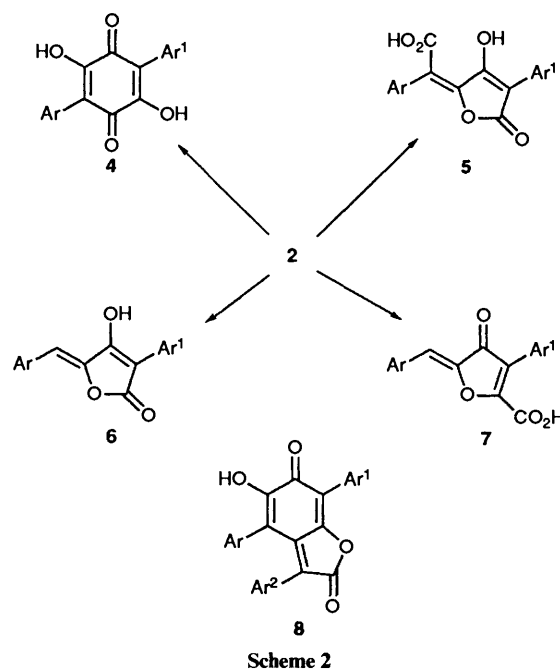
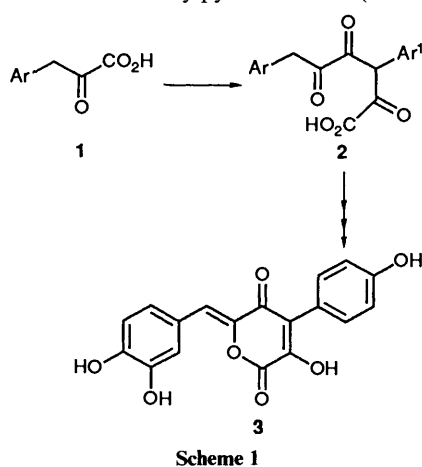
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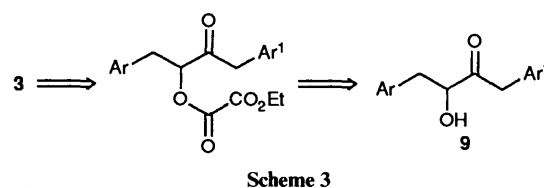
A synthesis of the grevillin group of pyrandione pigments *e.g.* **3**, **23** and **24** present in fungi is described. The synthesis, which is based on a biogenetic model, uses bis-benzylacyloins **9** and their corresponding oxalate derivatives as key intermediates (Scheme 3). Treatment of the grevillins **25–c** with sodium ethoxide in ethanol effects their quantitative isomerisation into the corresponding terphenylquinone pigments **4a–c**. Perkin-type condensations between the terphenylquinones **4** and arylacetic acids in the presence of sodium acetate–acetic anhydride then produces the xylerythrin pigments **29a–e**, whereas rearrangements of **4** in the presence of dimethyl sulphoxide leads to pulvinic acid derivatives, *e.g.* **31**, **32** and **5**. These synthetic studies interrelate the biosynthetic origins of the pigment types **3**, **4**, **5** and **8** together with the related pulvinone **6** and furanone **7** fungal pigments.

Grevillin is the generic name used to describe the group of orange and red pyrandione pigments *e.g.* **3** which have been isolated from fungi of the genus *Suillus*.¹ The grevillins co-occur with the related terphenylquinones **4** and pulvinic acids **5**, and they have their biogenetic origins in the dimer **2** derived from enzymatic conversion of arylpyruvic acid **1** (Scheme 1).² The

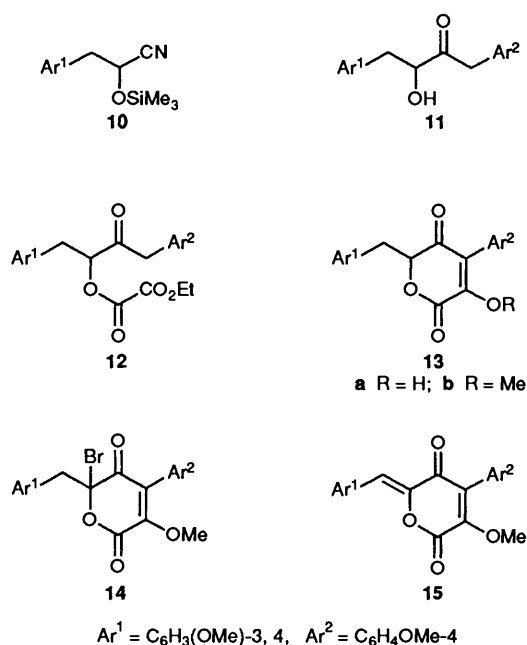


arylpyruvic acid dimer **2** also serves as the central intermediate in the biosynthesis of the terphenylquinones **4**, the pulvinic acids **5**, in addition to the pulvinone **6**, furanone **7** and xylerythrin **8** families of fungal pigments (Scheme 2).³ During investigations into the structure, the origins and the biogenetic interrelationships between the fungal pigment types **3–8**, we have earlier described concise syntheses of members of the pulvinone⁴ and the pulvinic acid⁵ groups of pigments. In continuation of these studies, we now describe a synthetic route to the grevillins **3** using benzylacyloins, *viz.* **9**, as key intermediates (Scheme 3),⁶ and illustrate their conversions *in vitro* into terphenylquinones **4** and thence the pulvinic acids **5**, and the xylerythrin group **8** of red quinone methide pigments found in the bark fungus *Peniophora sanguinea*.

Thus, using established methodology,⁷ addition of the Grignard reagent derived from 4-methoxybenzyl chloride to the *O*-trimethylsilyl cyanohydrin **10** first led to the unsymmetrically substituted benzylacyloin **11** in 71% yield. Addition of ethyl oxalyl chloride to the benzylacyloin **11** in the presence of



triethylamine next led to the corresponding oxalate **12**, which with 1,5-diazabicyclo[5.4.0]undec-5-ene (DBU) in dimethylformamide (DMF) at -15°C was smoothly converted into the dihydrogrevillin **13a**. The direct oxidation–dehydrogenation of **13a** to the corresponding grevillin proved problematic, *e.g.* 2,3-dichloro-5,6-dicyano-1,4-benzoquinone, Fremy's salt and palladium on charcoal led to recovered starting material or to intractable gums. Methylation of **13a** using diazomethane, however, proceeded smoothly producing the methyl ether **13b**, which when treated with bromine in acetic acid⁸ gave the



bromo derivative **14** in 77% yield. Elimination of the elements of hydrobromic acid from **14** in the presence of DBU, followed by demethylation of the resulting methyl grevillin **15** in the presence of boron tribromide⁹ finally produced grevillin B **3** as red crystals which showed spectroscopic data identical with those of natural material produced by the fruiting bodies of *Suillus grevillei*.¹⁰

In a similar series of reactions, the cyanohydrins **16a** and **16b** were elaborated to the corresponding grevillins **23** and **24** respectively (Scheme 4).

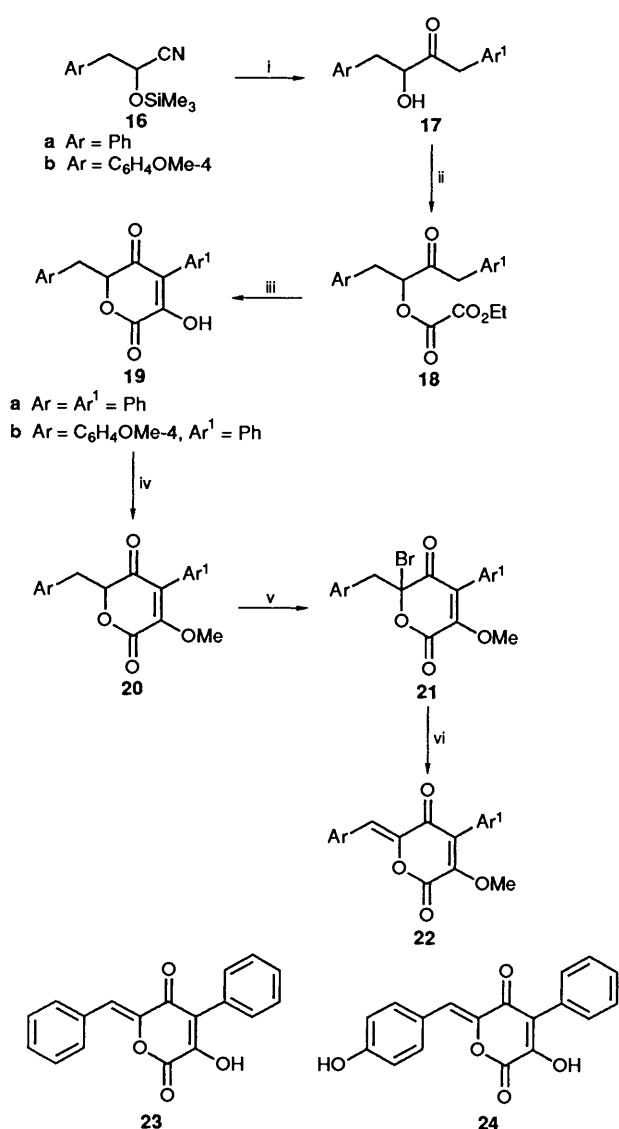
The isomeric relationships between the grevillin **25** and terphenylquinone **4** structures, which has also been examined by Steglich and co-workers,¹¹ is formally analogous to the relationship between the ylidenebutenolide and cyclopentenedione structures, *viz.* **26** and **27**, which we have examined in detail.¹² Indeed, this analogy could be fully demonstrated when the grevillin **25** was treated with sodium ethoxide in ethanol. A dense purple precipitate formed almost immediately, and acidification gave the known terphenylquinone **4a**¹³ in 90% yield. In a similar manner the synthetic grevillin derivatives **25b** and **25c** could be isomerised in the presence of sodium ethoxide to the terphenylquinones **4b** and **4c** respectively.

Naturally occurring terphenylquinones have previously been linked biogenetically to the xylerythrin **8** and the pulvinic acid **5** families of natural products.^{2,3} In our studies we have added further support for this biogenetic link with: (i) the conversion of the terphenylquinones **4a** and **4c** into the corresponding xylerythrins **29a-e** following Perkin reactions with arylacetic acids (to **28**)¹⁴ and demethylation in the presence of hydrobromic acid-acetic acid, and (ii) the conversions of **4a** and **4b** into the pulvinic acids **5a** and **5b/5c** respectively *via* the corresponding dilactone intermediates **30a** and **30b**,¹⁵ as outlined in Scheme 5.

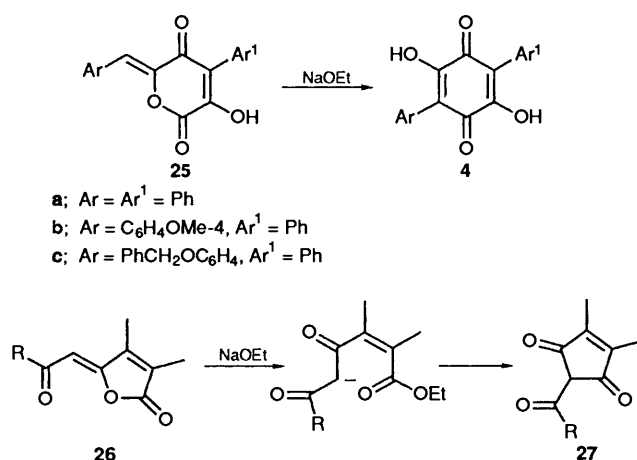
Experimental

For general experimental details see ref. 16. For NMR spectroscopic data *J* values are given in Hz. Ether refers to diethyl ether. Light petroleum refers to the fraction boiling in the range 60–80 °C.

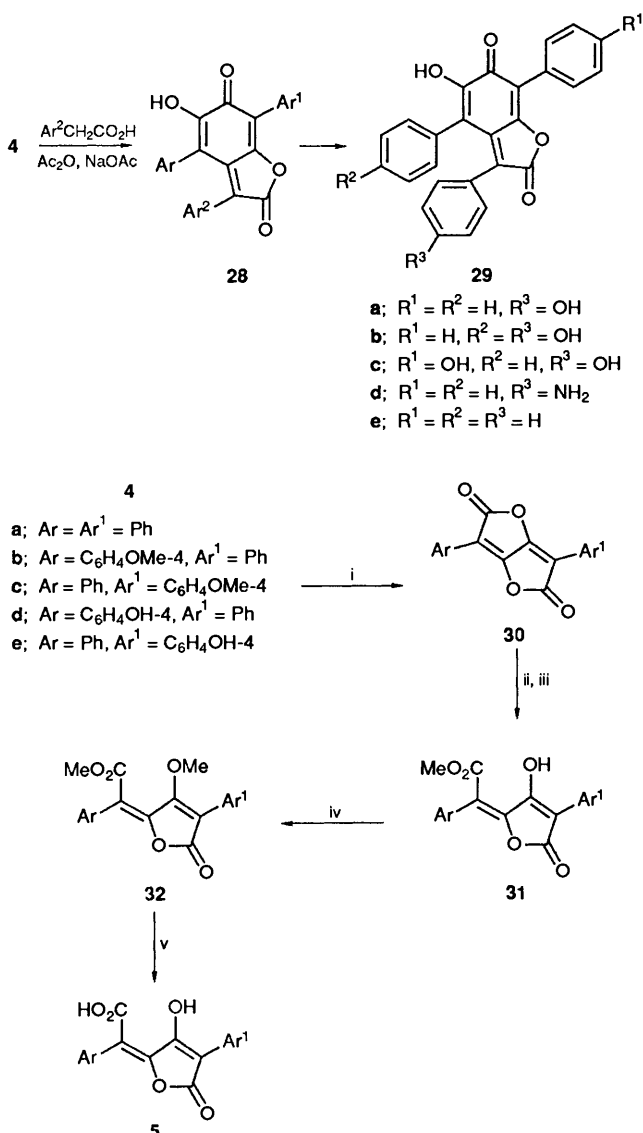
3-(3,4-Dimethoxyphenyl)-2-trimethylsilyloxypropanenitrile 10.—3,4-Dimethoxyphenylacetaldehyde (5.2 g) was added dropwise onto a stirred mixture of trimethylsilyl cyanide (3.6



Scheme 4 Reagents: i, PhCH₂MgCl; ii, CO₂EtCOCl, Et₃N; iii, DBU, DMF; iv, CH₂N₂; v, Br₂, AcOH; vi, DBU



cm³, 1 equiv.)¹⁷ and zinc iodide (1 crystal) at 0 °C under nitrogen. The mixture was stirred for a further 15 min and then distilled to give the title compound (4.85 g, 60%) as a colourless liquid, b.p. 140–145 °C at 2 mmHg, ν_{\max} (liquid film)/cm⁻¹ 2960m, 2250w, 1595w, 1515m, 1455m, 1425m, 1335w, 1260s, 1240s, 1160s, 1145s, 1105s, 1030s, 920m, 850s, 760m and 730s;



Scheme 5 Reagents: i, DMSO, Ac_2O , 100°C ; ii, NaOMe, MeOH; iii, c. HCl; iv, CH_2N_2 ; v, Me_3SiI

δ_{H} (no solvent) 0.00 (OTMS), 2.81 (d, J 7.0, CH_2), 3.65 ($2 \times \text{OMe}$), 4.35 (t, J 7.0, CH) and 6.65 ($3 \times \text{aryl}=\text{CH}$).

4-(3,4-Dimethoxyphenyl)-3-hydroxy-1-(4-methoxyphenyl)-butan-2-one 11.—A solution of compound **10** (1 g) in dry ether (10 cm^3) was added dropwise to a solution of 4-methoxybenzylmagnesium chloride (1.5 equiv.) in ether (50 cm^3) heated under reflux in an atmosphere of nitrogen. The mixture was heated under reflux for a further 3 h when a white precipitate formed. The mixture was allowed to cool to room temperature, and then poured onto ice-cooled dilute hydrochloric acid (20 cm^3). The two phase mixture was stirred vigorously at 25°C overnight. The mixture was extracted with ether ($2 \times 20 \text{ cm}^3$), and the combined ether extracts were dried and evaporated to leave a solid residue. Column chromatography then gave the title compound (0.84 g, 71%), which recrystallised from light petroleum as white needles, m.p. $81.5\text{--}82.5^\circ\text{C}$, $\nu_{\text{max}}(\text{CHCl}_3)/\text{cm}^{-1}$ 3550w, 2950m, 1720s, 1615s, 1530m, 1475s, 1305m, 1280m, 1160s and 865w; δ_{H} 2.80 (dd, J 6.8 and 14.4, 1 H), 3.10 (dd, J 4.7 and 14.4, 1 H), 3.16 (br, OH), 3.72 (CH_2), 3.78 (OMe), 3.83 (OMe), 3.86 (OMe), 4.47 (m, 1 H), 6.62–6.90 (m, $3 \times \text{aryl}=\text{CH}$), 6.84 (d, J 9.0, $2 \times \text{aryl}=\text{CH}$) and 7.04 (d, J 9.0, $2 \times \text{aryl}=\text{CH}$); δ_{C} 39.8, 44.9, 55.3, 55.8, 55.9, 76.7, 111.3, 112.5, 114.2, 121.3, 124.9,

128.9, 130.6, 148.1, 149.0, 158.8 and 209.5 (Found: C, 68.5; H, 6.7%; M^+ , 330.1471. $\text{C}_{19}\text{H}_{22}\text{O}_5$ requires C, 69.0; H, 6.7%; M , 330.1467).

1-(3,4-Dimethoxyphenyl)-4-(4-methoxyphenyl)-3-oxobutan-2-yl Ethyl Oxalate 12.—A solution of compound **11** (550 mg) and triethylamine (224 mm^3 , 1 equiv.) in dry tetrahydrofuran (THF) (25 cm^3) was added dropwise to a stirred solution of ethyl oxalyl chloride (235 mm^3 , 1 equiv.) in dry THF (25 cm^3) under nitrogen, whereupon a white precipitate formed immediately. The mixture was stirred at 25°C for 30 min, and then poured onto hydrochloric acid (2 mol dm^{-3} ; 30 cm^3). The mixture was extracted with ether ($2 \times 30 \text{ cm}^3$) and the combined ether extracts were dried and evaporated to leave a crude residue. Column chromatography gave the oxalate (416 mg, 58%) as a colourless oil, $\nu_{\text{max}}(\text{liquid film})/\text{cm}^{-1}$ 2950m, 1770s, 1750s, 1615w, 1515m, 1460m, 1300m, 1250s, 1185s, 1160s, 1030s, 910w, 860w and 760s; δ_{H} 1.37 (t, J 7.2, $\text{CO}_2\text{CH}_2\text{CH}_3$), 3.06 (d, J 6.3, CH_2), 3.63 (CH_2), 3.78 (OMe), 3.83 (OMe), 4.36 (q, J 7.2, $\text{CO}_2\text{CH}_2\text{CH}_3$), 5.36 (t, J 6.3, 1 H), 6.60–6.84 (m, $3 \times \text{aryl CH}$), 6.80 (d, J 9.0, $2 \times \text{aryl CH}$) and 7.05 (d, J 9.0, $2 \times \text{aryl CH}$); δ_{C} 13.9, 36.6, 46.1, 55.2, 55.8, 55.9, 63.4, 80.6, 111.2, 112.6, 114.2, 121.6, 124.3, 127.3, 127.3, 130.8, 148.3, 149.0, 156.9, 159.0 and 203.4 (Found: M^+ , 4.30.1629; $\text{C}_{23}\text{H}_{26}\text{O}_8$ requires M , 430.1627).

6-(3,4-Dimethoxybenzyl)-3-hydroxy-4-(4-methoxyphenyl)-pyran-2,5-dione 13a.—A solution of compound **12** (400 mg) in dry dimethylformamide (DMF) (10 cm^3) was added dropwise, over 15 min, to a stirred solution of 1,5-diazabicyclo[5.4.0]undec-5-ene (277 mm^3 , 2 equiv.) in dry DMF (10 cm^3) at -15°C under nitrogen. The resulting orange solution was stirred for 2 h and then poured onto hydrochloric acid (2 mol dm^{-3} ; 20 cm^3). The mixture was extracted with ether ($2 \times 20 \text{ cm}^3$), and the combined extracts were washed with water ($3 \times 20 \text{ cm}^3$), dried and evaporated to give the crude dione (252 mg, 71%) which recrystallised from light petroleum as a cream powder, m.p. $132\text{--}135^\circ\text{C}$, $\lambda_{\text{max}}(\text{EtOH})/\text{nm}$ 234 (ϵ 17 100), 253 (ϵ 13 000) and 334 (ϵ 5900); $\nu_{\text{max}}(\text{KBr})/\text{cm}^{-1}$ 3380br m, 2960w, 1730s, 1665s, 1615s, 1530s, 1475m, 1400w, 1365m, 1270s, 1215s, 1165m, 1035m, 840w and 785w; δ_{H} 3.33 (d, J 4.4, CH_2), 3.73 (OMe), 3.82 (OMe), 3.84 (OMe), 5.31 (t, J 4.4, CH), 6.60 (d, J 1.9, $1 \times \text{aryl}=\text{CH}$), 6.68 (dd, J 1.9 and 8.2, $1 \times \text{aryl CH}$), 6.75 (d, J 8.2, $1 \times \text{aryl}=\text{CH}$), 6.92 (d, J 8.9, $2 \times \text{aryl}=\text{CH}$) and 7.29 (d, J 8.9, $2 \times \text{aryl CH}$); δ_{C} 40.3, 55.3, 55.8, 55.9, 84.7, 111.2, 112.8, 113.6, 119.8, 121.8, 122.4, 125.9, 131.5, 148.5, 148.9, 160.1, 161.8 and 191.6 (Found: C, 65.4; H, 5.2%; M^+ , 384.1198. $\text{C}_{21}\text{H}_{20}\text{O}_7$ requires C, 65.6; H, 5.2%; M , 384.1209).

6-(3,4-Dimethoxybenzyl)-3-methoxy-4-(4-methoxyphenyl)-pyran-2,5-dione 13b.—An ethereal solution of diazomethane was added to a solution of compound **13a** (240 mg) in ether (10 cm^3) at 0°C , until the solution remained pale yellow. The mixture was stirred at 0°C for 5 min, and then evaporated to dryness to leave the dione **13b** (243 mg, 97%) which recrystallised from methanol as orange needles, m.p. $107\text{--}109^\circ\text{C}$, $\lambda_{\text{max}}(\text{EtOH})/\text{nm}$ 232 (ϵ 16 700), 250sh (ϵ 9040), 271sh (ϵ 6560) and 329 (ϵ 3380); $\nu_{\text{max}}(\text{CHCl}_3)/\text{cm}^{-1}$ 3400w, 2930m, 1720s, 1670s, 1595s, 1505m, 1445s, 1420m, 1350m, 1285s, 1250s, 1140s, 1025s, 860w, 830w, 805w, 760m and 730m; δ_{H} 3.30 (m, CH_2), 3.56 (OMe), 3.78 (OMe), 3.82 (OMe), 3.84 (OMe), 5.24 (dd, J 4.0 and 4.4, CH), 6.64 (d, J 1.9, $1 \times \text{aryl}=\text{CH}$), 6.70 (dd, J 1.9 and 8.1, $1 \times \text{aryl}=\text{CH}$), 6.80 (d, J 8.1, $1 \times \text{aryl}=\text{CH}$), 6.90 (d, J 8.9, $2 \times \text{aryl}=\text{CH}$) and 7.07 (d, J 8.9, $2 \times \text{aryl}=\text{CH}$); δ_{C} 40.5, 55.3, 55.8, 56.0, 61.1, 83.5, 111.3, 113.0, 113.5, 120.3, 122.4, 126.4, 129.5, 131.5, 148.4, 149.0, 153.6, 158.6, 160.2 and 193.3 (Found: M^+ , 398.1347. $\text{C}_{22}\text{H}_{22}\text{O}_7$ requires M , 398.1359).

6-Bromo-6-(3,4-dimethoxybenzyl)-3-methoxy-4-(4-methoxyphenyl)pyran-2,5-dione 14.—A solution of bromine in acetic acid (1% v/v solution; 0.65 cm³, 1 equiv.)⁸ was added dropwise over 15 min to a stirred solution of compound **13b** (50 mg) in acetic acid (2 cm³), under nitrogen. The mixture was stirred at room temperature for 2 h, diluted with water (10 cm³) and extracted with ether (2 × 20 cm³). The combined extracts were washed with water (6 × 20 cm³), dried and evaporated to leave the bromide **14** (46 mg, 77%) as an orange oil, λ_{\max} (EtOH)/nm 229 (ϵ 19 700), 254 (ϵ 13 000), 269 (ϵ 11 700) and 336 (ϵ 4 160); ν_{\max} (CHCl₃)/cm⁻¹ 2940m, 2840m, 1750s, 1685s, 1605s, 1495w, 1450m, 1335s, 1290s, 1140s, 1020m, 960w, 910w and 865w; δ_{H} 3.72 (d, *J* 14.2, 1 H), 3.82 (OMe), 3.84 (OMe), 3.85 (OMe), 3.86 (OMe), 3.95 (d, *J* 14.2, 1 H), 6.76 (d, *J* 8.1, 1 × aryl =CH), 6.84 (d, *J* 1.9, 1 × aryl =CH), 6.86 (dd, *J* 8.1 and 1.9, 1 × aryl =CH), 6.95 (d, *J* 8.9, 2 × aryl =CH) and 7.26 (d, *J* 8.9, 2 × aryl =CH); δ_{C} 46.1, 55.4, 55.9, 56.0, 61.5, 91.2, 111.1, 113.8, 114.3, 120.5, 123.9, 125.8, 127.8, 131.9, 148.6, 148.9, 150.4, 156.3, 160.6 and 185.9 (Found: M⁺, 478.0463 and 476.0460. C₂₂H₂₁BrO₇ requires *M*, 478.0450 and 476.0470). The bromide was used without further purification.

6-(3,4-Dimethoxybenzylidene)-3-methoxy-4-(4-methoxyphenyl)pyran-2,5-dione 15.—A solution of compound **14** (32 mg) in dry benzene (3 cm³) was added dropwise to a stirred solution of 1,5-diazabicyclo[5.4.0]undec-5-ene (10 mm³, 1 equiv.) in dry benzene (15 cm³) under nitrogen. After 10 min a black precipitate formed. The mixture was poured onto hydrochloric acid (2 mol cm⁻³; 10 cm³) and extracted with ether (2 × 20 cm³). The combined ether extracts were dried and evaporated to leave a residue which was purified by column chromatography to give the trimethylgrevillin (10.5 mg, 40%) as a solid which recrystallised from methanol as orange-yellow crystals, m.p. 175–176 °C (lit.,¹⁰ 177–179 °C), λ_{\max} (EtOH)/nm 230 (ϵ 14 000), 281 (ϵ 10 700), 336 (ϵ 12 000) and 410 (ϵ 5670); ν_{\max} (CHCl₃)/cm⁻¹ 2920m, 2830m, 1740s, 1680m, 1610s, 1590s, 1500w, 1460m, 1445m, 1425m, 1360s, 1340m, 1150s, 1100m, 1025m, 915s, 870w and 840w; δ_{H} 3.86 (OMe), 3.95 (2 × OMe), 3.97 (OMe), 6.93 (d, *J* 8.5, 1 × aryl =CH), 6.98 (d, *J* 8.9, 2 × aryl =CH), 7.04 (1 × =CH), 7.38 (d, *J* 8.9, 2 × aryl =CH), 7.50 (dd, *J* 2.0 and 8.5, 1 × aryl =CH) and 7.58 (d, *J* 2.0 H, 1 × aryl =CH); δ_{C} 55.3, 56.0, 61.3, 111.2, 113.6, 114.0, 120.3, 121.1, 125.1, 128.8, 131.8, 133.8, 143.1, 149.1, 150.9, 151.3, 155.4, 160.4 and 177.9 (Found: M⁺, 369.1209. C₂₂H₂₀O₇ requires *M*, 369.1209).

3-Hydroxy-6-(3,4-dihydroxybenzylidene)-4-(4-hydroxyphenyl)pyran-2,5-dione (Grevillin B) 3.—A solution of boron tribromide in hexane (1 mol dm⁻³; 177 mm³, 7 equiv.) was added dropwise to a stirred solution of compound **15** (10 mg) in dry dichloromethane (10 cm³) which was heated under reflux in a nitrogen atmosphere. The mixture was stirred and heated under reflux for a further 3.5 h, leading to a deep red precipitate. Concentrated hydrochloric acid¹⁰ (2 cm³) was added, and the mixture was then stirred vigorously for an additional 30 min. The mixture was poured onto water (15 cm³) and then extracted with ethyl acetate (2 × 30 cm³). The combined ethyl acetate extracts were washed successively with water (30 cm³), saturated aqueous mannitol (30 cm³) and water (30 cm³) and then dried and evaporated to leave a solid residue. Purification by column chromatography on acetylated polyamide gave the 'free' grevillin (3.3 mg, 38%), which recrystallised from ethanol as red crystals, m.p. 258 °C (decomp.) [lit.,^{1c} 275 °C (decomp.)/lit.,¹⁰ 350 °C with 250 °C (decomp.)]; λ_{\max} (EtOH)/nm 286 (ϵ 7250), 298sh (ϵ 6820) and 405 (ϵ 5650); ν_{\max} (KBr)/cm⁻¹ 3370br s, 3040s, 1725s, 1610m, 1580m, 1515m, 1450m, 1380m, 1300m, 1260s, 1230s, 1130m, 1045w, 1010w, 875w, 850w, 780w and 735w; δ_{H} ([²H₆]acetone) 6.88 (d, *J* 8.8, 2 × aryl =CH), 6.92 (1 × =CH), 6.92 (d, *J* 8.3, 1 × aryl =CH), 7.28 (dd, *J* 1.9 and

8.3, 1 × aryl =CH), 7.43 (d, *J* 8.8, 2 × aryl =CH) and 7.60 (d, *J* 1.9, 1 × aryl =CH) (Found: M⁺, 340.0579. C₁₈H₁₂O₇ requires *M*, 340.0579). The spectral data were identical with those reported by Edwards^{1c} and by Steglich¹⁰ and their collaborators for the natural material.

3-Phenyl-2-trimethylsilyloxypropanenitrile 16a.—Phenylacetaldehyde (13.6 g) was added dropwise to a stirred mixture of trimethylsilyl cyanide (15 cm³, 1 equiv.) and zinc iodide (1 crystal) at 0 °C under a nitrogen atmosphere. The mixture was stirred for a further 15 min and then distilled to give the title compound (24.6 g, 99%),¹⁷ as a colourless liquid, b.p. 90–94 °C at 0.5 mmHg, ν_{\max} (liquid film)/cm⁻¹ 2960br m, 1950w, 1720w, 1605w, 1500m, 1465m, 1365m, 1260s, 1115s, 940m, 880s, 860s, 765s and 705s; δ_{H} (no solvent) 0.00 (OTMS), 2.86 (d, *J* 6.3, CH₂), 4.42 (t, *J* 6.3, CH) and 7.15 (5 × aryl =CH).

3-Hydroxy-1,4-diphenylbutan-2-one 17a.—A solution of compound **16a** (14 g) in dry ether (100 cm³) was added dropwise to a solution of benzylmagnesium bromide (1.5 equiv.) in ether (400 cm³) under reflux in an atmosphere of nitrogen. The mixture was heated under reflux for a further 3 h when a white precipitate formed. The mixture was allowed to cool to room temperature, and then poured onto ice cooled dilute hydrochloric acid (300 cm³). The two-phase mixture was stirred vigorously at 25 °C overnight. The mixture was extracted with ether (2 × 200 cm³), and the combined ether extracts were then dried and evaporated to leave a solid residue. Purification by column chromatography gave the title compound (13.75 g, 90%), which recrystallised from light petroleum as white prismatic crystals, m.p. 56–58 °C (lit.,¹⁸ 59 °C), ν_{\max} (CHCl₃)/cm⁻¹ 3490m, 3020m, 2910m, 1705s, 1600w, 1490w, 1395m, 1325m, 1105m, 1050s and 910w; δ_{H} 2.80 (dd, *J* 7.3 and 14.1, CH), 3.10 (dd, *J* 4.8 and 14.1, CH), 3.31 (d, *J* 5.7, OH), 3.69 (CH₂), 4.40 (m, CH) and 7.00–7.40 (m, 10 × aryl =CH); δ_{C} 40.0, 45.4, 76.8, 126.7, 127.0, 128.4, 128.6, 129.3, 129.5, 133.3, 136.7 and 209.2 (Found: C, 80.15; H, 6.9%; M⁺, 240.1160. Calc. for C₁₆H₁₆O₂: C, 80.0; H, 6.7%; M, 240.1151).

3-Oxo-1,4-diphenylbutan-2-yl Ethyl Oxalate 18a.—A solution of compound **17a** (500 mg) and triethylamine (293 mm³, 1 equiv.) in dry THF (15 cm³) was added dropwise to a stirred solution of ethyl oxalyl chloride (235 mm³, 1 equiv.) in dry THF (10 cm³) under nitrogen, whereupon a white precipitate formed immediately. The mixture was stirred at 25 °C for 30 min, and then poured onto hydrochloric acid (2 mol dm⁻³; 30 cm³). The mixture was extracted with ether (2 × 30 cm³), and the combined ether extracts were then dried and evaporated to leave an oily residue. Column chromatography of the latter gave the oxalate (480 mg, 68%), as a colourless oil, ν_{\max} (liquid film)/cm⁻¹ 3000br w, 1765s, 1745s, 1600w, 1495w, 1455w, 1310m, 1185s, 1010m, 920w, 865w, 750m and 750s; δ_{H} 1.28 (t, *J* 6.5, CO₂CH₂CH₃), 3.10 (dd, *J* 5.4 and 7.2, CH₂), 3.69 (CH₂), 4.25 (q, *J* 6.5, CO₂CH₂CH₃), 5.45 (dd, *J* 5.4 and 7.2, CH) and 7.00–7.45 (m, 10 × aryl =CH); δ_{C} 13.9, 36.9, 46.9, 63.4, 80.4, 127.3, 127.4, 128.7, 129.5, 129.7, 132.4, 135.0, 156.8, 156.9 and 202.9 (Found: M⁺, 340.1315. C₂₀H₂₀O₅ requires *M*, 340.1310).

6-Benzyl-3-hydroxy-4-phenylpyran-2,5-dione 19a.—A solution of compound **18a** (4 g) in dry DMF (30 cm³) was added dropwise, over 15 min, to a stirred solution of 1,5-diazabicyclo[5.4.0]undec-5-ene (3.6 cm³, 2 equiv.) in dry DMF (100 cm³) at -15 °C under nitrogen. The resulting orange-red solution was stirred for a further 3 h and then poured onto hydrochloric acid (2 mol dm⁻³; 100 cm³). The mixture was extracted with ether (2 × 100 cm³), and the combined ether extracts were then washed with water (3 × 75 cm³), dried and evaporated to leave a cream solid. Recrystallisation of the latter

from heptane gave the *title compound* (3.14 g, 91%) as a white powder, m.p. 120–122 °C, $\lambda_{\max}(\text{EtOH})/\text{nm}$ 235 (ϵ 11 540) and 307 (ϵ 8120); $\nu_{\max}(\text{CHCl}_3)/\text{cm}^{-1}$ 3400s, 3040br m, 1720s, 1665s, 1600w, 1490w, 1360s, 1280s, 1180s, 1075m, 1040m, 950w and 885w; δ_{H} 3.38 (d, J 4.7, CH_2), 5.31 (t, J 4.7, CH) and 7.10–7.45 (m, $10 \times \text{aryl}=\text{CH}$); δ_{C} 40.4, 84.7, 127.6, 128.0, 128.7, 129.0, 133.7, 149.0, 161.4 and 191.0 (Found: C, 73.3; H, 4.9%; M^+ , 294.0877. $\text{C}_{18}\text{H}_{14}\text{O}_4$ requires: C, 73.5; H, 4.8%; M , 294.0892).

6-Benzyl-3-methoxy-4-phenylpyran-2,5-dione 20a.—An ethereal solution of diazomethane was added to a solution of compound **19a** (500 mg) in ether (30 cm^3) at 0 °C, until the solution remained pale yellow. The mixture was stirred at 0 °C for 15 min, and then evaporated to dryness to leave an oily residue. Column chromatography gave the *title compound* (429 mg, 82%), as a colourless oil, $\lambda_{\max}(\text{EtOH})/\text{nm}$ 228 (ϵ 6660), 248sh (ϵ 4150), 292 (ϵ 3130) and 302 (ϵ 3020); $\nu_{\max}(\text{liquid film})/\text{cm}^{-1}$ 3000br w, 1735s, 1675s, 1615m, 1600m, 1495w, 1445w, 1355m, 1330m, 1295m, 1210s, 1150s, 1080w, 1060w, 910w, 820w, 755s and 695s; δ_{H} 3.30 (d, J 4.5, CH_2), 3.44 (OMe), 5.18 (t, J 4.5, CH) and 6.90–7.40 (m, $10 \times \text{aryl}=\text{CH}$), δ_{C} 40.8, 61.3, 83.5, 127.6, 128.0, 128.5, 128.8, 129.1, 129.6, 130.0, 130.3, 134.0, 154.0, 158.2 and 192.6 (Found: M^+ , 308.1048. $\text{C}_{19}\text{H}_{16}\text{O}_4$ requires M , 308.1049).

6-Benzyl-6-bromo-3-methoxy-4-phenylpyran-2,5-dione 21a.—A solution of bromine in acetic acid (0.5% v/v solution; 12.9 cm^3 , 1 equiv.) was added dropwise over 15 min to a stirred solution of compound **20a** (400 mg) in acetic acid (12 cm^3), under nitrogen. The mixture was stirred at room temperature for 3 h and then diluted with water (50 cm^3) and extracted with ether (2 \times 50 cm^3). The combined ether extracts were washed with water (4 \times 30 cm^3), dried and evaporated to leave the *bromide* (452 mg, 90%), as a light yellow oil, $\lambda_{\max}(\text{EtOH})/\text{nm}$ 229 (ϵ 12 500), 291 (ϵ 6360) and 302 (ϵ 6070); $\nu_{\max}(\text{liquid film})/\text{cm}^{-1}$ 3000br m, 1750s, 1685s, 1610s, 1600s, 1495m, 1450s, 1430w, 1350s, 1310s, 1190s, 1130s, 1035w, 980w, 940m, 915m, 865m, 765s, 740s and 705s; δ_{H} 3.74 (d, J 14.1, CH), 3.81 (OMe), 4.00 (d, J 14.1, CH) and 7.15–7.50 (m, $10 \times \text{aryl}=\text{CH}$); δ_{C} 46.2, 61.6, 91.0, 127.8, 128.2, 128.5, 129.6, 130.2, 131.4, 133.3, 157.0 and 185.4 (3 carbons of low intensity not showing) (Found: M^+ , 388.0167 and 386.0131. $\text{C}_{19}\text{H}_{15}\text{BrO}_4$ requires M , 388.0133 and 386.0154). The bromide was used without further purification.

6-Benzylidene-3-methoxy-4-phenylpyran-2,5-dione 22a.—A solution of compound **21a** (420 mg) in dry benzene (10 cm^3) was added dropwise to a stirred solution of 1,5-diazabicyclo[5.4.0]undec-5-ene (178 mm^3 , 1.1 equiv.) in dry benzene (150 cm^3) under nitrogen. Initially a deep yellow solution was produced, which became cloudy and finally black within a few minutes. The mixture was poured onto hydrochloric acid (2 mol dm^{-3} ; 50 cm^3) and then extracted with ether (2 \times 100 cm^3). The combined ether extracts were dried and evaporated to leave an oily residue which was purified by column chromatography to give the methyl grevillin (75 mg, 22%), which recrystallised from heptane as yellow needles, m.p. 149–151 °C, $\lambda_{\max}(\text{EtOH})/\text{nm}$ 232 (ϵ 9380), 246sh (ϵ 8500), 311sh (ϵ 5230) and 320 (ϵ 5300); $\nu_{\max}(\text{CHCl}_3)/\text{cm}^{-1}$ 1745s, 1670m, 1610s, 1460w, 1320s, 1180s, 1120m and 925w; δ_{H} 3.93 (OMe), 7.05 (1 \times =CH), 7.22–7.52 (m, 8 \times aryl =CH) and 7.78–8.00 (m, 2 \times aryl =CH); δ_{C} 61.5, 120.0, 128.1, 129.0, 129.3, 130.2, 130.5, 131.9, 132.0, 144.0 and 178.0 (4 carbons of low intensity not showing) (Found: M^+ , 306.0886. $\text{C}_{19}\text{H}_{14}\text{O}_4$ requires M , 306.0890).

6-Benzylidene-3-hydroxy-4-phenylpyran-2,5-dione 23.—A solution of boron tribromide in hexane (1 mol dm^{-3} ; 660 mm^3 , 4 equiv.) was added dropwise to a stirred solution of compound

22a (50 mg) in dry dichloromethane (12 cm^3) which was heated under reflux in a nitrogen atmosphere. The mixture was stirred and heated under reflux for a further 30 min after which the resulting orange solution was poured onto methanol (20 cm^3), and diluted with water (20 cm^3). The mixture was extracted with ether (2 \times 30 cm^3), and the combined ether extracts were dried and evaporated to leave a solid residue. Column chromatography on acetylated polyamide gave the 'free' grevillin (44 mg, 92%), which recrystallised from ethanol as yellow plates, m.p. 233–235 °C (lit.,¹⁹ 237–238 °C), $\lambda_{\max}(\text{EtOH})/\text{nm}$ 201 (ϵ 16 400), 260 (ϵ 15 200) and 345 (ϵ 15 300); $\nu_{\max}(\text{KBr})/\text{cm}^{-1}$ 3300m, 1715s, 1590s, 1490w, 1445w, 1370s, 1210s, 995m, 760m, 735m and 695s; $\delta_{\text{H}}([\text{}^2\text{H}_6\text{]}\text{-DMSO})$ 6.98 (1 \times =CH), 7.30–7.60 (m, 8 \times aryl =CH) and 7.85–8.00 (m, 2 \times aryl =CH) (Found: C, 73.7; H, 4.2%; M^+ , 292.0731. Calc. for $\text{C}_{18}\text{H}_{12}\text{O}_4$: C, 74.0; H, 4.1%; M , 292.0732).

3-(4-Methoxyphenyl)-2-trimethylsilyloxypropanenitrile 16b.—4-Methoxyphenylacetaldehyde (2.75 g) was added dropwise onto a stirred mixture of trimethylsilyl cyanide (2.45 cm^3 , 1 equiv.) and zinc iodide (1 crystal) at 0 °C under a nitrogen atmosphere. The mixture was stirred for a further 2 h and then distilled to give the *title compound* (2.71 g, 59%), as a colourless liquid, b.p. 70–75 °C at 0.5 mmHg, $\nu_{\max}(\text{liquid film})/\text{cm}^{-1}$ 2960m, 1610m, 1580w, 1510s, 1460w, 1440w, 1350w, 1300m, 1265s, 1180m, 1115s, 1035m, 935w, 885m, 850s and 760m; δ_{H} (no solvent) 0.00 (OTMS), 2.80 (d, J 7.0, CH_2), 3.52 (OMe), 4.37 (t, J 7.0, CH), 6.70 (d, J 8.5, 2 \times aryl =CH) and 7.03 (d, J 8.5, 2 \times aryl =CH).

3-Hydroxy-4-(4-methoxyphenyl)-1-phenylbutan-2-one 17b.—A solution of compound **16b** (2.7 g) in dry ether (25 cm^3) was added dropwise to a refluxing ethereal solution (20 cm^3) of benzylmagnesium bromide (1.5 equiv.) under nitrogen. The mixture was stirred at ambient temperature overnight and then poured onto ice-cooled dilute hydrochloric acid (150 cm^3). The layers were separated and the organic phase was washed with brine (2 \times 20 cm^3), dried and evaporated to leave a solid residue. Purification of the latter by column chromatography gave the *title compound* (1.9 g, 65%) which recrystallised from light petroleum as white crystals, m.p. 74–76 °C, $\nu_{\max}(\text{CHCl}_3)/\text{cm}^{-1}$ 3480m, 2910m, 1710s, 1610s, 1585m, 1440m, 1300s, 1275m, 1110m, 1030m and 910w; δ_{H} 2.80 (dd, J 7.1 and 14.2, 1 H), 3.10 (dd, J 4.9 and 14.2, 1 H), 3.22 (br, OH), 3.75 (CH_2), 3.76 (OMe), 4.45 (br, m, CH), 6.82 (d, J 8.8, 2 \times aryl =CH) and 7.00–7.35 (m, 7 \times aryl =CH); δ_{C} 39.4, 45.8, 55.4, 77.1, 114.2, 127.4, 128.9, 129.6, 130.4, 130.9, 133.2, 158.8 and 209.2 (Found: C, 75.6; H, 6.9%; M^+ , 270.1241. $\text{C}_{17}\text{H}_{18}\text{O}_3$ requires C, 75.5; H, 6.7%; M , 270.1255).

Ethyl 1-(4-Methoxyphenyl)-4-phenyl-3-oxobutan-2-yl Oxalate 18b.—A solution of compound **17b** (1.25 g) and triethylamine (0.62 cm^3 , 1 equiv.) in dry THF (50 cm^3) was added dropwise to a stirred solution of ethyl oxalyl chloride (0.52 ml, 1 equiv.) in dry THF (50 cm^3), under nitrogen, whereupon a white precipitate formed immediately. The mixture was stirred at 25 °C for 30 min, and then poured onto hydrochloric acid (2 mol dm^{-3} ; 50 cm^3). The mixture was extracted with ether (2 \times 75 cm^3), and the combined ether extracts were then dried and evaporated to yield the *title compound* (1.68 g, 98%) as a light yellow oil, $\nu_{\max}(\text{liquid film})/\text{cm}^{-1}$ 2940br s, 1740br s, 1610w, 1510m, 1450m, 1300m, 1245s, 1180s, 1110w, 1040m, 825m, 730m and 695m; δ_{H} 1.38 (t, J 7.0 $\text{CO}_2\text{CH}_2\text{CH}_3$), 3.08 (d, J 6.3, CH_2), 3.70 (CH_2), 3.80 (OMe), 4.38 (q, J 7.0, $\text{CO}_2\text{CH}_2\text{CH}_3$), 5.37 (t, J 6.3, CH), 6.80 (d, J 9.0, 2 \times aryl =CH) and 7.00–7.40 (m, 7 \times aryl =CH); δ_{C} 13.9, 36.1, 47.0, 55.3, 63.4, 80.6, 114.1, 126.8, 127.3, 128.7, 129.7, 130.6, 132.4, 157.0, 158.9 and 203.1 (Found: M^+ , 370.1416. $\text{C}_{21}\text{H}_{22}\text{O}_6$ requires M , 370.1414).

3-Hydroxy-6-(4-methoxybenzyl)-4-phenylpyran-2,5-dione

19b.—A solution of compound **18b** (1.65 g) in dry DMF (50 cm³) was added dropwise, over 15 min, to a stirred solution of 1,5-diazabicyclo[5.4.0]undec-5-ene (1.30 cm³, 2 equiv.) in dry DMF (20 cm³) at -15 °C, under nitrogen. The mixture was stirred for 2 h and then quenched by addition of dilute hydrochloric acid (50 cm³). The mixture was extracted with ether (2 × 50 cm³), and the combined ether extracts were then washed with water (3 × 50 cm³), dried and evaporated to leave a crude residue. Column chromatography of the latter gave the *title compound* (1.06 g, 73%) as a pale yellow solid which recrystallised from heptane as yellow needles, m.p. 131–132 °C, $\lambda_{\max}(\text{EtOH})/\text{nm}$ 226 (ϵ 20 600), 285 (ϵ 7350) and 305 (ϵ 7650); $\nu_{\max}(\text{CHCl}_3)/\text{cm}^{-1}$ 3430m, 3000br w, 1730s, 1675s, 1615m, 1520w, 1365s, 1185w, 1120w, 955w, 890w and 845w; δ_{H} 3.34 (d, J 4.5, CH₂), 3.74 (OMe), 5.29 (t, J 4.5, CH), 6.78 (d, J 8.0, 2 × aryl =CH), 7.18–7.50 (m, 5 × aryl =CH) and 7.65 (d, J 8.0, 2 × aryl =CH); δ_{C} 39.7, 55.3, 85.0, 114.2, 122.2, 125.4, 127.7, 128.1, 129.2, 129.9, 131.2, 149.0, 159.2, 161.6 and 191.1 (Found: C, 70.4; H, 5.2%; M⁺, 324.0989. C₁₉H₁₆O₅ requires C, 70.4; H, 5.0%; M, 324.0996).

3-Methoxy-6-(4-methoxybenzyl)-4-phenylpyran-2,5-dione

20b.—An ethereal solution of diazomethane was added to a solution of compound **19b** (65 mg) in ether (3 cm³) at 0 °C, until the solution remained pale yellow. The mixture was stirred at 0 °C for 30 min, and then evaporated to dryness to leave the *title compound* (66 mg, 97%) as a viscous yellow oil, $\lambda_{\max}(\text{EtOH})/\text{nm}$ 225 (ϵ 18 000), 254 (ϵ 5800), 283 (ϵ 6540) and 294 (ϵ 6050); $\nu_{\max}(\text{liquid film})/\text{cm}^{-1}$ 3400br w, 2950br m, 1730s, 1675s, 1610s, 1600s, 1510s, 1445s, 1365s, 1300s, 1250s, 1205s, 1175s, 1145s, 1110m, 1060m, 1035s, 965w, 845m, 765m and 695m; δ_{H} 3.26 (d, J 4.0, CH₂), 3.52 (OMe), 3.73 (OMe), 5.18 (t, J 4.0, CH), 6.78 (d, J 8.0, 2 × aryl =CH), 7.04 (d, J 8.0, 2 × aryl =CH) and 6.93–7.40 (m, 5 × aryl =CH); δ_{C} 40.0, 55.3, 61.3, 83.7, 114.1, 125.9, 127.9, 128.0, 128.5, 129.1, 129.5, 130.0, 131.4, 159.1 and 192.8 (Found: M⁺, 338.1136. C₂₀H₁₈O₅ requires M, 338.1155).

6-Bromo-3-methoxy-6-(4-methoxybenzyl)-4-phenylpyran-2,5-dione

21b.—A solution of bromine in acetic acid (1% v/v solution; 0.81 cm³, 1 equiv.) was added dropwise over 15 min to a stirred solution of compound **20b** (54 mg) in acetic acid (2 cm³), under nitrogen. The mixture was stirred at ambient temperature for 2 h and then diluted with water (10 cm³) and extracted with ether (2 × 20 cm³). The combined ether extracts were washed with water (3 × 20 cm³), dried and evaporated to give the *title compound* (36 mg, 54%) as a very unstable yellow oil, $\nu_{\max}(\text{CHCl}_3)/\text{cm}^{-1}$ 2950br m, 1765s, 1695s, 1615s, 1605s, 1495m, 1450m, 1350s, 1310s, 1135s, 1035m 990w, 940w, 875w and 850w; δ_{H} 3.75 (OMe), 3.82 (OMe), 3.40–3.75 (m, CH₂), 6.80 (d, J 8.7, 2 × aryl =CH) and 7.10–7.55 (m, 7 × aryl =CH). The bromide was used without further purification.

3-Methoxy-6-(4-methoxybenzylidene)-4-phenylpyran-2,5-dione

22b.—A solution of compound **21b** (98 mg) in dry benzene (10 cm³) was added dropwise to a stirred solution of 1,5-diazabicyclo[5.4.0]undec-5-ene (35 mm³, 1 equiv.) in dry benzene (60 cm³) under nitrogen. After 45 min the mixture was poured onto dilute hydrochloric acid (25 cm³) and extracted with ether (2 × 50 cm³). The combined ether extracts were dried and evaporated to leave an oily residue which was purified by column chromatography to give the *title compound* (15 mg, 20%), which recrystallised from methanol as yellow needles, m.p. 162–163 °C, $\lambda_{\max}(\text{EtOH})/\text{nm}$ 234 (ϵ 8610), 267 (ϵ 10 850) and 402 (ϵ 8080); $\nu_{\max}(\text{CHCl}_3)$ 2940w, 1740s, 1665w, 1590s, 1510w, 1365m, 1315m, 1165s, 1110w, 1030w and 915w; δ_{H} 3.86 (OMe), 3.93 (OMe), 6.96 (d, J 8.9, 2 × aryl =CH), 7.04 (1 × =CH), 7.41 (5 × aryl =CH) and 7.90 (d, J 8.9,

2 × aryl =CH); δ_{C} 55.4; 61.4, 114.6, 120.3, 124.8, 128.1, 129.1, 129.2, 130.2, 133.7, 134.1, 142.9, 151.4, 155.4, 161.6 and 177.6 (Found: M⁺, 336.1005. C₂₀H₁₆O₅ requires M, 336.0998).

3-Hydroxy-6-(4-hydroxybenzylidene)-4-phenylpyran-2,5-dione

24.—A solution of boron tribromide in hexane (1 mol dm⁻³; 255 mm³, 5 equiv.) was added dropwise to a stirred solution of compound **22** (14 mg) in dry dichloromethane (10 cm³) which was heated under reflux in a nitrogen atmosphere. The mixture was stirred and heated under reflux for a further 4 h after which concentrated hydrochloric acid¹⁰ (3 cm³) was added to it and the mixture was stirred vigorously for an additional 1 h. The mixture was poured onto water (15 cm³) and extracted with ethyl acetate (2 × 30 cm³). The combined ethyl acetate extracts were washed successively with water (30 cm³), saturated aqueous mannitol (30 cm³) and water (30 cm³) and then dried and evaporated to leave a solid residue. Purification of this by column chromatography on acetylated polyamide gave the 'free' *grevillin* (5.4 mg, 42%), which recrystallised from ethanol as yellow–orange crystals, m.p. 240–248 °C (decomp.), $\lambda_{\max}(\text{EtOH})/\text{nm}$ 264 (ϵ 13 400), 283 (ϵ 12 700) and 400 (ϵ 11 700); $\nu_{\max}(\text{KBr})/\text{cm}^{-1}$ 3400br s, 1640m, 1575s, 1515m, 1450w, 1380s, 1345m, 1245m, 1210s, 1175s, 1005w, 845w, 770w, 740w, and 700w; $\delta_{\text{H}}([\text{C}_6\text{H}_6]\text{acetone})$ 6.98 (d, J 8.7, 2 × aryl =CH), 7.01 (1 × =CH), 7.30–7.48 (m, 3 × aryl =CH), 7.50–7.60 (m, 2 × aryl =CH) and 7.88 (d, J 8.7, 2 × aryl =CH); $\delta_{\text{C}}([\text{C}_6\text{H}_6]\text{acetone})$ 116.8, 119.4, 124.9, 125.7, 128.4, 129.2, 131.3, 134.7, 144.1, 150.0, 158.3, 160.4 and 177.5 (Found: M⁺, 308.0676. C₁₈H₁₂O₅ requires M, 308.0683).

3,6-Dihydroxy-2,5-diphenyl-1,4-benzoquinone (Polyporic Acid)

4a.—A solution of compound **23** (50 mg) in dry ethanol (5 cm³) was treated with sodium ethoxide (0.17 mol dm⁻³ solution; 1 cm³, 1 equiv.).¹¹ A deep purple precipitate formed immediately and the mixture was stirred, at room temperature, under nitrogen for 30 min. The reaction was acidified with hydrochloric acid (2 mol dm⁻³; 5 cm³) and extracted with ether (2 × 20 cm³). The ether extracts were dried and evaporated to give polyporic acid (45 mg, 90%), which recrystallised from acetone as dark brown needles, m.p. 300–301 °C (lit.,¹³ 303–305 °C); $\lambda_{\max}(\text{EtOH})/\text{nm}$ 262 (ϵ 21 300), 325 (ϵ 5330) and 475 (ϵ 254); $\nu_{\max}(\text{KBr})/\text{cm}^{-1}$ 3300s, 1610s, 1595sh s, 1325s, 1310s, 1250s, 1000s, 770m, 725m and 695m, $\delta_{\text{H}}([\text{C}_6\text{H}_6]\text{-DMSO})$ 7.38 (10 × aryl =CH) (Found: M⁺, 292.0721. C₁₈H₁₂O₄ requires M, 292.0732). The sample was identical with a specimen synthesised by the alternative route described by Shildneck and Adams.^{13,20}

6-(4-Benzyloxybenzylidene)-3-hydroxy-4-phenylpyran-2,5-dione

25c.—The *grevillin* which was prepared in 64% yield according to the method described by Steglich,¹⁰ had m.p. 246–248 °C (lit.,¹⁰ m.p. 247–249 °C), $\lambda_{\max}(\text{EtOH})/\text{nm}$ 238sh (ϵ 11 350), 256 (ϵ 12 560), 285sh (ϵ 9150), 362sh (ϵ 13 110) and 383 (ϵ 14 350); $\nu_{\max}(\text{KBr})/\text{cm}^{-1}$ 3300m, 1720s, 1640w, 1580s, 1500m, 1375s, 1250s, 1200s, 1170s, 995m, 830m, 730m and 695m; $\delta_{\text{H}}([\text{C}_6\text{H}_6]\text{-DMSO})$ 5.21 (CH₂), 6.99 (1 × =CH), 7.17 (d, J 8.8, 2 × aryl =CH), 7.30–7.60 (m, 10 × aryl =CH) and 7.93 (d, J 8.8, 2 × aryl =CH) (Found: C, 75.5; H, 4.6%; M⁺, 398.1148. Calc. for C₂₅H₁₈O₅: C, 75.4; H, 4.6%; M, 398.1149).

3,6-Dihydroxy-2-(4-methoxyphenyl)-5-phenyl-1,4-benzoquinone

4b.—A suspension of compound **25b** (50 mg) in dry methanol (3 cm³) was treated with sodium methoxide (2.6 mol dm⁻³ solution; 1.6 cm³, 2.6 equiv.), whereupon a deep purple precipitate formed almost immediately. The mixture was stirred for 30 min under nitrogen and then acidified with hydrochloric acid (2 mol dm⁻³, 5 cm³). The green precipitate which formed was collected by vacuum filtration to give the

terphenylquinone (41 mg, 82%), which recrystallised from ethanol as greenish black microcrystalline square plates, m.p. 272–274 °C (lit.,¹¹ m.p. 280 °C), $\lambda_{\max}(\text{EtOH})/\text{nm}$ 267 (ϵ 22 520), 350 (ϵ 3513), 497 (ϵ 500); $\nu_{\max}(\text{KBr})/\text{cm}^{-1}$ 3320s, 1615s, 1515m, 1450w, 1330s, 1315s, 1260s, 1190m, 1040m, 1005s, 845m, 810m, 775m, 735m and 705m; $\delta_{\text{H}}([\text{}^2\text{H}_6\text{J}]\text{-DMSO})$ 3.79 (OMe), 6.97 (d, J 8.9, 2 \times aryl =CH), 7.35 (d, J 8.9, 2 \times aryl =CH) and 7.30–7.40 (m, 5 \times aryl =CH) (Found: M^+ , 322.0844. $\text{C}_{19}\text{H}_{14}\text{O}_5$ requires M , 322.0840).

3-Methoxy-6-(4-hydroxybenzylidene)-4-phenylpyran-2,5-dione 25b.—A solution of boron tribromide in hexane (1 mol dm^{-3} ; 67 mm^3 , 5 equiv.) was added dropwise to a stirred solution of compound **22b** (4.5 mg) in dry dichloromethane (5 cm^3), which was heated under reflux in a nitrogen atmosphere. The mixture was stirred and heated under reflux for a further 30 min after which concentrated hydrochloric acid (1 cm^3) was added, and the mixture was stirred vigorously for an additional 30 min. The mixture was poured onto water (10 cm^3) and then extracted with ethyl acetate (2 \times 20 cm^3). The combined ethyl acetate extracts were washed successively with water (20 cm^3), saturated aqueous mannitol (20 cm^3) and water (20 cm^3) and then dried and evaporated to leave a solid residue. Purification of the latter by column chromatography on acetylated polyamide gave the 'free' *grevillin* (2 mg, 46%), which recrystallised from methanol as orange crystals, m.p. 216–218 °C, $\lambda_{\max}(\text{EtOH})/\text{nm}$ 244sh (ϵ 13 500), 260 (ϵ 16 800), 280sh (ϵ 13 600) and 387 (ϵ 13 900); $\nu_{\max}(\text{KBr})/\text{cm}^{-1}$ 3320m, 1725s, 1655m, 1590s, 1515m, 1430w, 1380s, 1310s, 1265s, 1210s, 1185s, 1100w, 1035m, 1015w, 840m, 760m and 705m; $\delta_{\text{H}}([\text{}^2\text{H}_6\text{J}]\text{-acetone})$ 3.89 (OMe), 7.03 (1 \times =CH), 7.07 (d, J 8.9, 2 \times aryl =CH), 7.30–7.60 (m, 5 \times aryl =CH) and 7.94 (d, J 8.9, 2 \times aryl =CH); $\delta_{\text{C}}([\text{}^2\text{H}_6\text{J}]\text{-acetone})$ 55.8, 115.3, 119.0, 125.6, 125.8, 128.4, 129.2, 130.9, 131.3, 134.3, 144.3, 150.1, 158.2, 162.2 and 177.5 (Found: M^+ , 322.0846. $\text{C}_{19}\text{H}_{14}\text{O}_5$ requires M 322.0841).

2-(4-Benzoyloxyphenyl)-3,6-dihydroxy-5-phenyl-1,4-benzoquinone 4c.—A solution of compound **25c** (97 mg) in dry ethanol (3 cm^3) was treated with sodium ethoxide (2.6 mol dm^{-3} solution; 2 ml, 2 equiv.). A deep purple precipitate formed immediately and the mixture was stirred, at room temperature, under nitrogen for 30 min. The reaction was acidified with hydrochloric acid (2 mol dm^{-3} ; 5 cm^3) to give a green precipitate, which was collected by vacuum filtration and dried to give the terphenylquinone (89 mg, 92%), which recrystallised from ethanol as black plates, m.p. 268–270 °C (lit.,¹¹ m.p. 278 °C), $\lambda_{\max}(\text{EtOH})/\text{nm}$ 270 (ϵ 49 000), 349 (ϵ 3710) and 494 (ϵ 340); $\nu_{\max}(\text{KBr})/\text{cm}^{-1}$ 3300s, 1605br s, 1510m, 1320s, 1250s, 1180m, 995s, 835w, 800w, 765w, 720w and 695w; $\delta_{\text{H}}([\text{}^2\text{H}_6\text{J}]\text{-DMSO})$ 5.15 (CH_2), 7.04 (d, J 9.0, 2 \times aryl =CH) and 7.25–7.55 (m, 12 \times aryl =CH) (Found: C, 75.7; H, 4.8%; M^+ , 98.1145. Calc. for $\text{C}_{25}\text{H}_{18}\text{O}_5$: C, 75.4; H, 4.6%; M , 398.1149).

5-Hydroxy-3-(4-hydroxyphenyl)-4,7-diphenylbenzofuran-2,6-dione (Xylerythrin) 29a.—A solution of polyporic acid **4a** (50 mg) and 4-hydroxyphenylacetic acid (26 mg) in acetic anhydride (1 cm^3), containing an excess of sodium acetate (5 equiv.) was heated under reflux for 3 h. The acetic anhydride was decomposed by the addition of water (1 cm^3) after which a mixture of 48% hydrogen bromide (1 cm^3) and acetic acid (1 cm^3) was added. The mixture was heated under reflux for 0.5 h and then cooled to room temperature and diluted with water (15 cm^3). The mixture was extracted with chloroform (3 \times 15 cm^3), and the combined chloroform extracts were then dried and evaporated to leave a solid residue. Purification by column chromatography gave xylerythrin (27 mg, 39%), as black crystals with a green lustre, m.p. 263–267 °C (CHCl_3) (lit.,¹⁴ m.p. 265–268 °C, lit.,²¹ m.p. 253–255 °C), $\lambda_{\max}(\text{EtOH})/\text{nm}$ 254

(ϵ 26 600), 359 (ϵ 9300) and 465 (ϵ 14 050); $\nu_{\max}(\text{KBr})/\text{cm}^{-1}$ 3600–3100br m, 1770m, 1740w, 1615m, 1600s, 1500w, 1400m, 1325m, 1270m, 1150m, 1015m, 835w, 750m and 695m; $\delta_{\text{H}}([\text{}^2\text{H}_6\text{J}]\text{-acetone})$ 6.49 (d, J 8.9, 2 \times aryl =CH), 6.96 (d, J 8.9, 2 \times aryl =CH), 7.05–7.15 (m, 3 \times aryl =CH), 7.15–7.30 (m, 2 \times aryl =CH), 7.40–7.55 (m, 3 \times aryl =CH) and 7.60–7.70 (m, 2 \times aryl =CH); $\delta_{\text{C}}([\text{}^2\text{H}_6\text{J}]\text{-acetone})$ 115.0, 115.3, 121.0, 128.3, 128.4, 128.8, 129.3, 130.6, 130.8, 131.2, 131.5, 132.6, 132.8, 133.7, 137.8, 148.5, 158.5, 159.6, 168.2 and 181.4 (Found: M^+ , 408.1006. $\text{C}_{26}\text{H}_{16}\text{O}_5$ requires M , 408.0993). Identical spectral data were recorded for an authentic sample of natural xylerythrin.¹⁴

5-Hydroxy-3,4-bis-(hydroxyphenyl)-7-phenylbenzofuran-2,6-dione (Peniophorin) 29b and 5-Hydroxy-3,7-bis-(4-hydroxyphenyl)-4-phenylbenzofuran-2,6-dione (Isopeniophorin) 29c.—A solution of compound **4c** (40 mg) and 4-hydroxyphenylacetic acid (1 equiv.) in acetic anhydride (1 cm^3), containing an excess of sodium acetate (5.25 equiv.) was heated under reflux for 3 h. The acetic acid was decomposed by the addition of water (1 cm^3), and then a mixture of 48% hydrogen bromide (1 cm^3) and acetic acid (1 cm^3). The mixture was heated under reflux for 0.5 h and then cooled to room temperature and diluted with water (15 cm^3). The mixture was extracted with chloroform (2 \times 15 cm^3), and the combined organic extracts were then dried, and evaporated to give a solid residue containing both isopeniophorin and peniophorin. The isomers were separated by column chromatography to give: (i) isopeniophorin (eluted first) (9 mg, 22%), m.p. 298–307 °C (decomp.), ($\text{CHCl}_3\text{-AcOEt}$); $\lambda_{\max}(\text{EtOH})/\text{nm}$ 263 (ϵ 20 000), 302sh (ϵ 8600), 400sh (ϵ 4400) and 470 (ϵ 9500); $\nu_{\max}(\text{KBr})/\text{cm}^{-1}$ 3470s, 3330s, 2930w, 1765s, 1600s, 1595s, 1505w, 1405m, 1330s, 1280m, 1225m, 1180m, 1150m, 1030m, 845w and 700w; $\delta_{\text{H}}([\text{}^2\text{H}_6\text{J}]\text{-acetone})$ 6.50 (d, J 8.6, 2 \times aryl =CH), 6.90–7.00 (m, 4 \times aryl =CH), 7.05–7.15 (m, 3 \times aryl =CH), 7.17–7.25 (m, 2 \times aryl =CH) and 7.53 (d, J 8.7, 2 \times aryl =CH); $\delta_{\text{C}}([\text{}^2\text{H}_6\text{J}]\text{-acetone})$ 115.2, 115.7, 121.1, 121.5, 128.2, 128.3, 128.8, 129.2, 130.8, 131.6, 132.5, 132.7, 133.0, 138.1, 148.5, 157.8, 158.5, 168.5 and 181.8 (Found: M^+ , 424.0959. $\text{C}_{26}\text{H}_{16}\text{O}_6$ requires M , 424.0942) and (ii) peniophorin (eluted second) (4.5 mg, 11%), m.p. 287–293 °C (decomp.) (CHCl_3) [lit.,²² m.p. 300–305 °C (decomp.)]; $\lambda_{\max}(\text{EtOH})/\text{nm}$ 268 (ϵ 23 400), 302sh (ϵ 7980), 389sh (ϵ 6500), 410sh (ϵ 7100) and 468 (ϵ 8780); $\nu_{\max}(\text{KBr})/\text{cm}^{-1}$ 3550m, 3300m, 1775s, 1695m, 1625s, 1600s, 1400w, 1330m, 1100w, 1010w, 980w, 890w and 830w; $\delta_{\text{H}}([\text{}^2\text{H}_6\text{J}]\text{-acetone})$ 6.56 (d, J 8.8, 2 \times aryl =CH), 6.58 (d, J 8.7, 2 \times aryl =CH), 7.01 (d, J 8.8, 2 \times aryl =CH), 7.06 (d, J 8.7, 2 \times aryl =CH), 7.45–7.55 (m, 3 \times aryl =CH) and 7.60–7.68 (m, 2 \times aryl =CH); $\delta_{\text{C}}([\text{}^2\text{H}_6\text{J}]\text{-acetone})$ 115.2, 121.1, 123.4, 128.3, 128.8, 129.3, 130.0, 130.4, 130.7, 131.1, 131.5, 132.0, 132.7, 133.0, 138.1, 148.1, 157.9, 159.6 and 168.3 (Found: M^+ , 424.0958. $\text{C}_{26}\text{H}_{16}\text{O}_6$ requires M , 424.0942). Identical spectral data were recorded for an authentic sample of natural peniophorin.²²

3-(4-Aminophenyl)-5-hydroxy-4,7-diphenylbenzofuran-2,6-dione 29d.—A solution of polyporic acid (50 mg) and 4-amino-phenylacetic acid (1.2 equiv.) in acetic anhydride (1 cm^3) containing an excess of sodium acetate (5 equiv.) was heated under reflux for 3 h. The acetic anhydride was decomposed by addition of water (1 cm^3), and then a mixture of 48% hydrobromic acid (1 cm^3) and acetic acid (1 cm^3) was added. The mixture was heated under reflux for 0.5 h and then cooled to room temperature and diluted with water (15 cm^3). The mixture was extracted with chloroform (2 \times 15 cm^3), and the combined organic extracts were then dried and evaporated to leave the aminoxylerythrin (6 mg, 10%), which recrystallised from chloroform as blue crystals, m.p. 233–240 °C (decomp.); $\lambda_{\max}(\text{EtOH})/\text{nm}$ 252sh (ϵ 18 300), 258 (ϵ 18 500), 359 (ϵ 6410) and 561 (ϵ 6940); $\nu_{\max}(\text{KBr})/\text{cm}^{-1}$ 3400br s, 1780m, 1625s, 1605s,

1515m, 1450w, 1415m, 1340m, 1315m, 1195w, 1165m, 1030m, 900w, 840w, 815w, 765w, 705m and 625w; $\delta_{\text{H}}([\text{}^2\text{H}_6\text{]}-\text{acetone})$ 6.30 (d, *J* 8.8, 2 \times aryl =CH), 6.90 (d, *J* 8.8, 2 \times aryl =CH), 7.15–7.25 (m, 3 \times aryl =CH), 7.25–7.35 (m, 2 \times aryl =CH), 7.40–7.55 (m, 3 \times aryl =CH) and 7.60–7.70 (m, 2 \times aryl =CH); $\delta_{\text{C}}([\text{}^2\text{H}_6\text{]}-\text{acetone})$ 113.6, 117.8, 119.5, 128.3, 128.5, 128.8, 129.2, 130.8, 131.5, 131.8, 133.2, 134.8, 138.0, 148.2, 151.4, 168.4 and 181.3 (2 carbons low intensity not showing) (Found: M^+ , 407.1149. $\text{C}_{26}\text{H}_{17}\text{NO}_4$ requires *M*, 407.1155).

5-Hydroxy-3,4,7-triphenylbenzofuran-2,6-dione (Deoxyxylerythrin) 29e—A solution of polyporic acid (50 mg) and phenylacetic acid (1 equiv.) in acetic anhydride (1 cm^3) containing an excess of sodium acetate (5 equiv.) was heated under reflux for 3 h. The acetic anhydride was decomposed by addition of water (1 cm^3), and then a mixture of 48% hydrobromic acid (1 cm^3) and acetic acid (1 cm^3) was added. The mixture was heated under reflux for 0.5 h, cooled to room temperature and then diluted with water (15 cm^3). The mixture was extracted with chloroform (2 \times 15 cm^3) and the combined organic extracts were then dried and evaporated to leave a solid residue.

Purification by column chromatography gave the deoxyxylerythrin (13 mg, 20%) as red crystals, m.p. 215–220 °C (decomp.) (CHCl_3); $\lambda_{\text{max}}(\text{EtOH})/\text{nm}$ 251 (ϵ 20 700), 282sh, 391sh, 396 (ϵ 12 700) and 476 (ϵ 1110); $\nu_{\text{max}}(\text{KBr})/\text{cm}^{-1}$ 3600–3200br s, 1775s, 1630s, 1610s, 1445w, 1410w, 1335m, 1310m, 1200w, 1150w, 1025w, 935w, 785w, 755m and 700m; $\delta_{\text{H}}([\text{}^2\text{H}_6\text{]}-\text{acetone})$ 7.00–7.10 (m, 5 \times aryl =CH), 7.10–7.40 (m, 5 \times aryl =CH), 7.45–7.60 (m, 3 \times aryl =CH) and 7.60–7.70 (m, 2 \times aryl =CH); $\delta_{\text{C}}([\text{}^2\text{H}_6\text{]}-\text{acetone})$ 114.9, 115.4, 126.2, 128.1, 128.3, 128.4, 128.9, 129.4, 129.6, 130.7, 130.8, 131.5, 140.4, 148.8, 158.3, 167.8 and 181.6 (Found: M^+ , 392.1031. $\text{C}_{26}\text{H}_{16}\text{O}_4$ requires *M*, 392.1044).

3,6-Diphenylfuro[3,2-b]furan-2,5-dione (Pulvinic Anhydride) 30a—A solution of polyporic acid (50 mg) in dimethyl sulphoxide (2 cm^3) and acetic anhydride (1 cm^3) was warmed to 100 °C in a nitrogen atmosphere for 15 min.¹⁵ The cooled mixture was diluted with water (10 cm^3), and then extracted with chloroform (2 \times 30 cm^3). The combined chloroform extracts were washed successively with brine (2 \times 30 cm^3) and saturated aqueous sodium hydrogencarbonate (20 cm^3) and then dried and evaporated to leave the dilactone (44 mg, 89%), which recrystallised from chloroform as yellow microcrystalline plates, m.p. 226–228 °C (lit.,¹⁵ m.p. 221–222 °C); $\lambda_{\text{max}}(\text{EtOH})/\text{nm}$ 233 (ϵ 9600), 288sh (ϵ 6170) and 373 (ϵ 10 200); $\nu_{\text{max}}(\text{KBr})/\text{cm}^{-1}$ 1820s, 1795sh s, 1660s, 1495m, 1450m, 1365s, 1340s, 1320m, 1195w, 1165m, 1055m, 1005w, 920w, 875s, 800s, 780s, 730m, 690s, and 660m; $\delta_{\text{H}}([\text{}^2\text{H}_6\text{]}-\text{DMSO})$ 7.10–7.40 (m, 8 \times aryl =CH) and 8.12 (dd, *J* 8.4 and 1.2, 2 \times aryl =CH); $\delta_{\text{C}}([\text{}^2\text{H}_6\text{]}-\text{DMSO})$ 95.0, 117.5, 121.1, 125.2, 126.8, 127.1, 127.8, 130.0, 132.3, 135.8, 152.3, 166.9, 168.3 and 170.6 (Found: C, 74.1; H, 3.6%; M^+ , 290.0564. Calc. for $\text{C}_{18}\text{H}_{10}\text{O}_4$: C, 74.5; H, 3.5%; *M*, 290.0579).

(E)-4-Hydroxy-5-(α -methoxycarbonylbenzylidene)-3-phenylfuran-2-(5H)-one (Vulpinic Acid) 31a—Aqueous sodium hydroxide (18 mol dm^{-3} ; 0.9 cm^3) was added to a suspension of pulvinic anhydride (1.7 g) in methanol (25 cm^3), and after 5 min the resulting clear solution was diluted with water (25 cm^3) and acidified with concentrated hydrochloric acid. The precipitate was filtered off, and then placed under high vacuum for several hours to leave vulpinic acid (1.79 g, 95%) as a solid which recrystallised from methanol as yellow square plates, m.p. 150–151 °C (lit.,²³ m.p. 148 °C); $\lambda_{\text{max}}(\text{EtOH})/\text{nm}$ 234 (ϵ 11 986), 276 (ϵ 10 318), 366 (ϵ 11 106); $\nu_{\text{max}}(\text{CHCl}_3)/\text{cm}^{-1}$ 3000br w, 2640br w, 1780s, 1680m, 1615s, 1605s, 1435m, 1315s, 1280s, 1065m, 965m and 905m; δ_{H} 3.86 (CO_2Me), 7.20–7.60 (m, 8 \times aryl =CH), 8.10–8.30 (m, 2 \times aryl =CH) and 13.87 (OH); δ_{C}

54.5, 105.2, 115.8, 127.9, 128.1, 128.3, 128.4, 128.6, 128.9, 129.9, 131.9, 154.9, 160.2, 165.9 and 171.7 (Found: C, 70.8; H, 4.3%; M^+ , 322.0832. Calc. for $\text{C}_{19}\text{H}_{14}\text{O}_5$: C, 70.8; H, 4.4%; *M*, 322.0839).

(E)-4-Methoxy-5-(α -methoxycarbonylbenzylidene)-3-phenylfuran-2-(5H)-one (Permethylylated Pulvinic Acid) 32a—An ethereal solution of diazomethane was added to a suspension of vulpinic acid (1.7 g) in ether (85 cm^3) at 0 °C, until the solution remained pale yellow. The solution was stirred at 0 °C for 3 h after which it was evaporated to leave permethylylated pulvinic acid (1.52 g, 86%), as a solid which recrystallised from methanol as cream rods, m.p. 140–141 °C (lit.,²⁴ m.p. 142–143 °C), $\lambda_{\text{max}}(\text{CHCl}_3)/\text{cm}^{-1}$ 224 (ϵ 7940), 233 (ϵ 11 613) and 331 (ϵ 22 129); $\nu_{\text{max}}(\text{CHCl}_3)/\text{cm}^{-1}$ 3000br w, 1770s, 1730s, 1630s, 1600m, 1490w, 1440w, 1370m, 1325m, 1300m, 1160m, 1130w, 1030w, 980m, 930m; δ_{H} 3.74 (CO_2Me), 3.85 (OMe), 7.25–7.50 (m, 8 \times aryl =CH) and 7.59–7.75 (m, 2 \times aryl =CH); δ_{C} 52.8, 61.3, 108.2, 116.4, 128.4, 128.5, 128.8, 129.0, 129.1, 129.4, 130.0, 131.0, 141.4, 162.6, 167.1 and 167.8 (Found: C, 71.3; H, 4.8%; M^+ , 336.1001. Calc. for $\text{C}_{20}\text{H}_{16}\text{O}_5$: C, 71.4; H, 4.8%; *M*, 336.0996).

(E)-5-(α -Carboxybenzylidene)-4-hydroxy-3-phenylfuran-2-(5H)-one (Pulvinic Acid) 5a—A deuteriochloroform solution of permethylylated pulvinic acid (32 mg) was treated with iodo-trimethylsilane (68 cm^3 , 5 equiv.)⁵ in a ^1H NMR tube, sealed under nitrogen and warmed at 55 °C. The reaction was monitored by ^1H NMR spectroscopy, following the disappearance of the methyl group signals, with concomitant formation of methyl iodide. After 3 d the persilylated pulvinic acid was hydrolysed with methanol (6 cm^3), and the mixture was then evaporated to give the crude product. Column chromatography gave the pulvinic acid (18 mg, 61%), which recrystallised from chloroform as orange elongated plates, m.p. 202–208 °C (lit.,²⁵ m.p. 202–207 °C, lit.,²⁶ m.p. 216–217 °C), $\lambda_{\text{max}}(\text{EtOH})/\text{nm}$ 241 (ϵ 13 127), 253 (ϵ 12 596) and 362 (ϵ 12 152); $\nu_{\text{max}}(\text{KBr})/\text{cm}^{-1}$ 3450m, 3000br w, 2500br w, 1750s, 1680m, 1620s, 1595s, 1500w, 1450m, 1375m, 1305w, 1270w, 1230w, 1170w, 1080w, 1065w, 970m, 920w, 790m, 730m and 700m; $\delta_{\text{H}}([\text{}^2\text{H}_6\text{]}-\text{acetone})$ 7.30–7.55 (m, 8 \times aryl =CH) and 8.05–8.20 (m, 2 \times aryl =CH); $\delta_{\text{C}}([\text{}^2\text{H}_6\text{]}-\text{acetone})$ 104.3, 117.5, 128.3, 128.7, 128.9, 129.1, 129.2, 130.3, 131.0, 134.2, 155.6, 162.2, 166.6 and 174.0 (Found: M^+ , 308.0677. $\text{C}_{18}\text{H}_{12}\text{O}_5$ requires *M*, 308.0685).

6-(4-Methoxyphenyl)-3-phenylfuro[3,2-b]furan-2,5-dione (4-Methoxypulvinic Acid Dilactone) 30b—A solution of compound 4b (26 mg) in dimethyl sulphoxide (0.4 cm^3) and acetic anhydride (0.2 cm^3) was warmed to 100 °C in a nitrogen atmosphere for 15 min. The mixture was cooled in an ice-bath and then the orange–yellow precipitate was filtered off. The solid was washed with water (5 cm^3) and then dried *in vacuo* to leave the dilactone (18 mg, 62%), which recrystallised from benzene as orange microcrystalline plates, m.p. 201–203 °C (lit.,²⁷ m.p. 200–201 °C), $\lambda_{\text{max}}(\text{CHCl}_3)/\text{nm}$ 243 (ϵ 18 416) and 410 (ϵ 22 865); $\nu_{\text{max}}(\text{CHCl}_3)/\text{cm}^{-1}$ 3000br w, 1825s, 1790sh s, 1670s, 1610s, 1500w, 1365m, 1345m, 1320m, 1305m, 1165m, 1030w, 875s, 840m and 660m; δ_{H} 3.87 (OMe), 7.00 (d, *J* 9.0, 2 \times aryl =CH), 7.40–7.55 (m, 3 \times aryl =CH) and 7.90–8.05 (m, 4 \times aryl =CH); δ_{C} 55.4, 100.9, 101.6, 114.7, 119.0, 126.6, 128.0, 129.1, 129.9, 155.1, 157.3, 161.0, 166.0 and 166.1 (Found: C, 71.3; H, 3.6%; M^+ , 320.0687. Calc. for $\text{C}_{19}\text{H}_{12}\text{O}_5$: C, 71.25; H, 3.8%; *M*, 320.0683).

(E)-4-Hydroxy-5-(α -methoxycarbonylbenzylidene)-3-(4-methoxyphenyl)furan-2-(5H)-one (Pinastric Acid) 31c and (E)-4-Hydroxy-5-(α -methoxycarbonyl-4-methoxybenzylidene)-3-phenylfuran-2-(5H)-one (Isopinastric Acid) 31b—Aqueous sodium hydroxide (18 mol dm^{-3} solution; 0.92 cm^3 , 2.8 equiv.) was

added to a suspension of 4-methoxypulvinic dilactone (1.9 g) in methanol (25 cm³), and after 5 min the resulting clear solution was diluted with water (25 cm³) and acidified with concentrated hydrochloric acid. The resulting precipitate was filtered off to give a 4:1 mixture (1.98 g, 95%) of the two pulvinates **31b** and **31c**. The mixture of isomers was boiled in methanol, and the insoluble pinastric acid **31c** was filtered off. It recrystallised from benzene as rectangular plates, m.p. 207–209 °C (lit.,²⁸ m.p. 202–204 °C), $\lambda_{\max}(\text{EtOH})/\text{nm}$ 208, 290 and 400; $\nu_{\max}(\text{CHCl}_3)/\text{cm}^{-1}$ 3000w, 2620w, 1775m, 1675m, 1600s, 1435w, 1110s, 1030w and 965m; δ_{H} 3.84 (OMe), 3.87 (CO₂Me), 6.96 (d, *J* 9.2, 2 × aryl =CH), 7.23–7.45 (m, 5 × aryl =CH), 8.12 (d, *J* 9.2, 2 × aryl =CH) and 13.50 (OH) (Found: C, 68.6; H, 4.5%; M⁺, 352.0931. Calc. for C₂₀H₁₆O₆: C, 68.2; H, 4.6%; M, 352.0942). The filtrate was evaporated to dryness and the residue was crystallised from methanol to give isopinastric acid **31b**, as microneedles, m.p. 122–124 °C (lit.,²⁸ m.p. 127–129 °C), $\lambda_{\max}(\text{EtOH})/\text{nm}$ 210, 229, 273 and 375; $\nu_{\max}(\text{CHCl}_3)/\text{cm}^{-1}$ 2960w, 2620w, 1770s, 1680s, 1430m, 1320s, 1050m, 965s and 910m; δ_{H} 3.85 (OMe), 3.89 (CO₂Me), 6.93 (d, *J* 9.2, 2 × aryl =CH), 7.10–7.45 (m, 5 × aryl =CH), 8.00–8.20 (m, 2 × aryl =CH) and 13.67 (OH) (Found: C, 68.3; H, 4.4%; M⁺, 352.0947. Calc. for C₂₀H₁₆O₆: C, 68.2; H, 4.6%; M, 352.0942).

(E)-4-Methoxy-5-(α -methoxycarbonyl-4-methoxybenzylidene)-3-phenylfuran-2(5H)-one (O-Methylisopinastric Acid) **32c**.—An ethereal solution of diazomethane was added to a suspension of isopinastric acid (400 mg) in ether (40 cm³) at 0 °C, until the ether solution remained pale yellow. The solution was stirred at 0 °C for 3 h and the solvent was then removed to leave O-methylisopinastric acid (426 mg, 95%) which recrystallised from methanol as yellow needles, m.p. 171–173 °C (lit.²⁸ m.p. 172–175 °C), $\lambda_{\max}(\text{CHCl}_3)/\text{nm}$ 358 (ϵ 24 600); $\nu_{\max}(\text{CHCl}_3)/\text{cm}^{-1}$ 2940w, 1760s, 1725s, 1630s, 1600s, 1370m, 1290m, 1160m, 985m and 940m; δ_{H} 3.76 (OMe), 3.84 (OMe), 3.89 (CO₂Me), 6.93 (d, *J* 8.9, 2 × aryl =CH), 7.40–7.70 (m, 5 × aryl =CH) and 7.65 (d, *J* 8.9, 2 × aryl =CH) (Found: C, 69.1; H, 5.0%; M⁺, 366.1141. Calc. for C₂₁H₁₈O₆: C, 68.8; H, 4.95%; M, 366.1098).

(E)-4-Methoxy-5-(α -methoxycarbonylbenzylidene)-3-(4-methoxyphenyl)furan-2(5H)-one [O-Methylpinastric Acid] **32b**.—An ethereal solution of diazomethane was added to a suspension of pinastric acid (100 mg) in ether (10 cm³) at 0 °C, until the ether solution remained pale yellow. The solution was stirred at 0 °C for 3 h and then evaporated to leave O-methylpinastric acid (106 mg, 95%), which recrystallised from benzene as pale yellow microcrystals, m.p. 140–142 °C (lit.,⁵ m.p. 140–141 °C), $\lambda_{\max}(\text{CHCl}_3)/\text{nm}$ 342 (ϵ 19 400); $\nu_{\max}(\text{CHCl}_3)/\text{cm}^{-1}$ 2940w, 1770s, 1730s, 1635m, 1605s, 1300m, 1160m and 930m; δ_{H} 3.79 (OMe), 3.84 (OMe), 3.90 (CO₂Me), 6.96 (d, *J* 9.2, 2 × aryl =CH), 7.30–7.45 (m, 3 × aryl =CH), 7.50 (d, *J* 9.2, 2 × aryl =CH) and 7.60–7.80 (m, 2 × aryl =CH) (Found: C, 68.7; H, 5.0%; M⁺, 366.1135. Calc. for C₂₁H₁₈O₆: C, 68.8; H, 4.95%; M, 366.1098).

(E)-5-(α -Carboxy-4-hydroxybenzylidene)-4-hydroxy-3-phenylfuran-2(5H)-one (4'-Hydroxypulvinic Acid) **5d**.—A solution of O-methylisopinastric acid (74 mg) in deuteriochloroform (1 cm³) was treated with iodotrimethylsilane (173 mm³, 6 equiv.) in a ¹H NMR tube, sealed under nitrogen and warmed to 55 °C. The reaction was monitored by ¹H NMR spectroscopy and after 3 d the resulting persilylated pulvinic acid was hydrolysed with methanol (6 cm³). The solvent was evaporated to leave a solid residue which was purified by column chromatography to give 4'-hydroxypulvinic acid (22 mg, 34%), m.p. 95–99 °C (resolidify 105 °C, remelt 266–289 °C), $\lambda_{\max}(\text{EtOH})/\text{nm}$ 244 (ϵ 12 300), 263 (ϵ 14 500) and 367 (ϵ 9500); $\lambda_{\max}(\text{EtOH} + 1$

drop NaOH)/nm 257 (ϵ 13 100), 267 (ϵ 12 600) and 383 (ϵ 20 400); $\nu_{\max}(\text{KBr})/\text{cm}^{-1}$ 3400br m, 3000br w, 2550br w, 1745m, 1675w, 1605s, 1595s, 1510w, 1445m, 1370m, 1265m, 1060m, 965m, 920w, 785m and 690m; $\delta_{\text{H}}([\text{}^2\text{H}_6\text{]}_{\text{acetone}})$ 6.90 (d, *J* 8.9, 2 × aryl =CH), 7.29 (d, *J* 8.9, 2 × aryl =CH), 7.20–7.45 (m, 3 × aryl =CH) and 8.06–8.17 (m, 2 × aryl =CH) (Found: M⁺, 324.0562. C₁₈H₁₂O₆ requires M, 324.0630). An accurate combustion analysis could not be obtained for this compound.

(E)-5-(α -Carboxybenzylidene)-4-hydroxy-3-(4-hydroxyphenyl)furan-2(5H)-one (4-Hydroxypulvinic Acid) **5e**.—A solution of pinastric acid (29 mg) in deuteriochloroform (1 cm³) was treated with iodotrimethylsilane (68 cm³, 6 equiv.) in a ¹H NMR tube, sealed under nitrogen and warmed to 55 °C. The reaction was conveniently monitored by ¹H NMR spectroscopy and after 3 d the resulting persilylated derivative was hydrolysed with methanol (6 cm³). The solvent was evaporated to leave a crude residue which was purified by column chromatography to give the pulvinic acid (12 mg, 35%), m.p. 85–90 °C, $\lambda_{\max}(\text{EtOH})/\text{nm}$ 256 (ϵ 13 600), 286sh and 397 (ϵ 6900); $\lambda_{\max}(\text{EtOH} + 1 \text{ drop NaOH})/\text{nm}$ 298 (ϵ 19 700) and 418 (ϵ 5900); $\nu_{\max}(\text{KBr})/\text{cm}^{-1}$ 3400br m, 300br m, 2600br w, 1740m, 1675m, 1600s, 1365m, 1260m, 1180m, 1060m, 960m, 840m, 735m and 710m; $\delta_{\text{H}}([\text{}^2\text{H}_6\text{]}_{\text{acetone}})$ 6.91 (d, *J* 9.0, 2 × aryl =CH), 7.42 (5 × =CH) and 8.03 (d, *J* 9.0, 2 × aryl =CH) (Found: M⁺, 324.0589. C₁₈H₁₂O₆ requires M, 324.0630). An accurate combustion analysis could not be obtained for this compound.

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References

- (a) W. Steglich, H. Besl and A. Prox, *Tetrahedron Lett.*, 1972, 4895; (b) H. Besl, I. Michler, R. Preuss and W. Steglich, *Z. Naturforsch., Teil C*, 1974, **29**, 784; (c) R. L. Edwards and M. Gill, *J. Chem. Soc., Perkin Trans. 1*, 1973, 1921.
- For a recent review see: M. Gill and W. Steglich, *Fortsch. Chem. Org. Naturst.*, 1987, **51**, 1.
- For a review see: G. Pattenden, *Fortsch. Chem. Org. Naturst.*, 1978, **35**, 133.
- (a) D. W. Knight and G. Pattenden, *J. Chem. Soc., Chem. Commun.*, 1975, 876; (b) D. W. Knight and G. Pattenden, *J. Chem. Soc., Chem. Commun.*, 1976, 635; (c) M. J. Begley, D. W. Knight and G. Pattenden, *Tetrahedron Lett.*, 1976, 131.
- (a) D. W. Knight and G. Pattenden, *J. Chem. Soc., Perkin Trans. 1*, 1979, 84; (b) G. Pattenden, N. Pegg and A. G. Smith, *Tetrahedron Lett.*, 1986, **27**, 403.
- Preliminary publication: G. Pattenden, N. A. Pegg and R. W. Kenyon, *Tetrahedron Lett.*, 1987, **28**, 4749. For related studies see: M. Gill and M. J. Kiefel, *Tetrahedron Lett.*, 1988, **29**, 2085; M. Gill, M. J. Kiefel, D. A. Lally and A. Ten, *Aust. J. Chem.*, 1990, **43**, 1497.
- P. Ruggli and P. Zeller, *Helv. Chim. Acta*, 1945, **28**, 741; L. R. Krepski, S. M. Heilmann and J. K. Rasmussen, *Tetrahedron Lett.*, 1983, **24**, 4075; M. Gill, M. J. Kiefel and D. A. Lally, *Tetrahedron Lett.*, 1986, **27**, 1933.
- J. Sedgeworth and G. R. Proctor, *J. Chem. Soc., Perkin Trans. 1*, 1985, 2677.
- J. F. W. McOmie, M. L. Watts and D. E. West, *Tetrahedron*, 1968, **24**, 2289.
- For an alternative synthesis of grevillins see H.-J. Lohrisch, L. Kopanski, R. Herrmann, H. Schmidt and W. Steglich, *Leibigs Ann. Chem.*, 1986, 177.
- H.-J. Lohrisch, H. Schmidt and W. Steglich, *Liebigs Ann. Chem.*, 1986, 195.
- N. G. Clemo, D. R. Gedge and G. Pattenden, *J. Chem. Soc., Perkin*

- Trans. 1*, 1981, 1448; cf. H-H. Lee, Y-T. Que and S. Ng, *J. Chem. Soc., Perkin Trans. 1*, 1985, 453.
- 13 P. R. Shildneck and R. Adams, *J. Am. Chem. Soc.*, 1931, **53**, 2373.
- 14 J. Gripenberg and J. Martikkala, *Acta Chem. Scand.*, 1969, **23**, 2583.
- 15 R. J. Wikholm and H. W. Moore, *J. Am. Chem. Soc.*, 1972, **94**, 6152.
- 16 P. Patel and G. Pattenden, *J. Chem. Soc., Perkin Trans. 1*, 1991, 1941.
- 17 D. A. Evans, L. K. Truesdale and G. L. Carroll, *J. Chem. Soc., Chem. Commun.*, 1973, 55.
- 18 P. Ruggli and P. Zeller, *Helv. Chem. Acta*, 1945, **28**, 741.
- 19 H-J. Lohrisch and W. Steglich, *Tetrahedron Lett.*, 1975, 2905.
- 20 N. A. Pegg, Ph.D. Thesis, University of Nottingham, 1987.
- 21 H-W. Wanzlick and U. Jahnke, *Chem. Ber.*, 1968, **101**, 3753.
- 22 J. Gripenberg and J. Martikkala, *Acta Chem. Scand.*, 1970, **24**, 3444.
- 23 A. Spiegel, *Justus Liebigs Ann. Chem.*, 1883, **219**, 1.
- 24 A. Schonberg and A. Sina, *J. Chem. Soc.*, 1946, 601.
- 25 R. Ramage, G. J. Griffiths and J. N. A. Sweeney, *J. Chem. Soc., Perkin Trans. 1*, 1984, 1547.
- 26 W. Zopf, *Justus Liebigs Ann. Chem.*, 1895, **284**, 107.
- 27 B. Akermark, *Acta Chem. Scand.*, 1961, **15**, 1695.
- 28 S. Agarwal and T. Seshadri, *Ind. J. Chem.*, 1964, **2**, 17.

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